Bow Valley College

Course Outline

CHEM1902

General Chemistry: Change and Equilibrium
Fall 2025 - Current

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CHEM1902 General Chemistry: Change and Equilibrium

COURSE DESCRIPTION

An introduction to university chemistry from theoretical and practical perspectives that focuses on an exploration of the fundamental links between kinetics, equilibria, and thermodynamics and explores acidity/basicity and redox behaviour using inorganic and organic examples.

REQUISITES	Complete the following courses: • CHEM1901 - General Chemistry: Structure and Bonding (4)
EQUIVALENTS	None
CREDITS	4
HOURS	60
ELIGIBLE FOR PLAR	No
ZERO TEXTBOOK COST	Yes

COURSE LEARNING OUTCOMES

Bow Valley College is committed to ensuring our graduates can demonstrate their abilities in key areas that will make them effective citizens and encourage their development as lifelong learners. In addition to the discipline-specific skills that learners acquire in their programs, the College has identified ten learning outcomes.

College-Wide Outcomes:

1. Communication
2. Thinking Skills
3. Numeracy and Financial Literacy
4. Working with Others
5. Digital Literacy
6. Positive Attitudes and Behaviours
7. Continuous Learning
8. Health and Wellness Awareness
9. Citizenship and Intercultural Competence
10. Environmental Sustainability

#	COURSE LEARNING OUTCOME(S)	COLLEGE WIDE OUTCOMES SUPPORTED
1	Explain the connections between temperature, kinetic energy, and reactivity using the kinetic molecular theory of ideal gases.	1, 2, 4, 5, 7, 8, and 10
2	Utilize the principles of chemical equilibria to forecast the degree of changes in aqueous chemistry, including acid/base reactions, dissociation of ionic compounds, and redox reactions in electrochemical cells.	1, 2, 4, 5, 7, 8, and 10
3	Determine the factors influencing reaction rates, represent these rates using graphs and symbols, and elucidate the molecular basis for these rates.	1, 2, 4, 5, 7, 8, and 10
4	Assess the changes in thermodynamic enthalpy and entropy during a chemical reaction to predict the potential for spontaneous occurrence, and explain methods to modify this spontaneity.	1, 2, 4, 5, 7, 8, and 10
5	Apply chemical equations and empirical data to address quantitative issues related to kinetics, thermodynamics, and equilibrium concepts.	1, 2, 4, 5, 7, 8, and 10
6	Accurate experimental reporting and effectively communicate the outcomes of chemical transformations through observable macroscopic results, molecular-scale models/representations, and mathematical expressions.	1, 2, 4, 5, 7, 8, and 10

COURSE MODULES AND SCHEDULE

**Course schedule subject to change, depending on delivery mode and term of study. For exact dates, please consult the Course Offering Information in Brightspace.*

WEEK/HOURS	MODULES
Week 1-2	Review of background topics: Stoichiometry of reactions; safety in the Chemistry lab
Week 3	Gases
Week 4-5	Chemical kinetics
Week 6-7	Chemical equilibrium
Week 8	Acid-Base equilibria
Week 9	Solubility
Week 10-11	Thermodynamics
Week 12	Electrochemistry
Week 13	Exam prep and course review
Week 14	Final exam

ASSESSMENT

COURSE LEARNING OUTCOME(S)	ASSESSMENT	WEIGHT
1, 2, 3, 4, 5	Learning activities	24%
1, 2, 3, 4, 5, 6	Laboratory component	16%
1, 2, 3	Midterm exam(s)	25%
1, 2, 3, 4, 5	Final exam	35%

Important: For details on each assignment and exam, please see the Course Offering Information.

PERFORMANCE STANDARDS

A minimum grade of D is required to pass this course. However, a program may require a higher grade in this course to progress in the program or to meet specific program completion requirements.

Please consult with the program area or contact the program chair for further details. A minimum Grade Point Average of 2.0 is required for graduation.

GRADING SCHEME

Grade	Percentage	Grade Point	Description
A+	95-100	4.0	Exceptional: superior knowledge of subject matter
A	90-94	4.0	Excellent: outstanding knowledge of subject matter
A-	85-89	3.67	
B+	80-84	3.33	
B	75-79	3.0	Very Good: knowledge of subject matter generally mastered
B-	70-74	2.67	
C+	67-69	2.33	
C	64-66	2.0	Satisfactory/Acceptable: knowledge of subject matter adequately mastered
C-	60-63	1.67	
D+	57-59	1.33	
D	50-56	1.0	Minimal Pass
F	Less than 50	0.0	Fail: an unsatisfactory performance

REQUIRED LEARNING RESOURCES

Required materials:

- The recommended textbook is the University of Calgary Department of Chemistry Textbook: https://chem-textbook.ucalgary.ca/. It is a free online resource.
- TopHat will be used for in-class responses. Use the link https://app.tophat.com/register/ to register for TopHat.
- A lab coat, a pair of goggles, and a hard-bound lab notebook are required for labs
- Non-programmable scientific calculator (Casio FX260, FX300, or similar)

Additional learning resources may be found in the Course Offering Information or in Brightspace.

ADDITIONAL INFORMATION

Additional information may be found in the Course Offering Information or in Brightspace.

ACADEMIC ACCOMMODATIONS

Learners with a disability (learning, physical, and/or mental health) may qualify for academic and exam accommodations. For more information, or to apply for accommodations, learners should make an appointment with Accessibility Services in the Learner Success Services (LSS) Department. Accessibility Services can also assist learners who may be struggling with learning but do not have a formal diagnosis. To make an appointment visit LSS on the first floor of the south campus or call 403-410-1440. It is the learner's responsibility to contact Accessibility Services and request academic accommodations. For more information, please visit our website at http://www.bowvalleycollege.ca/accessibility.

INSTITUTIONAL POLICIES

Bow Valley College is committed to the highest standards of academic integrity and honesty. Learners are urged to become familiar with and uphold the following policies: Academic Integrity (500-1-7), Learner Code of Conduct, Procedures and Guidelines (500-1-1), Learner Appeals (500-1-12), Attendance (500-1-10), Grading (500-1-6), Academic Continuance and Graduation (500-1-5), and Electronic Communications (300-2-13). Audio or video recording of lectures, labs, seminars, or any other teaching and learning environment by learners is allowed only with consent of the instructor as part of an approved accommodation plan. Recorded material is to be used solely for personal study and is not being used or distributed without prior written consent from the instructor.

Turnitin:

Students may be required to submit their course work to Turnitin, a third-party service provider engaged by BVC. Turnitin identifies plagiarism by checking databases of electronic books and articles, archived webpages, and previously submitted student papers. Students acknowledge that any course work or essays submitted to Turnitin will be included as source documents in the Turnitin.com reference database, where it will be used solely to detect plagiarism. The terms that apply to a student's use of Turnitin are described on Turnitin.com.

Online Exam Proctoring:

Examinations for this course may require proctoring through an online proctoring service. Online proctoring enables online exam taking within a controlled and monitored environment, thereby enhancing academic integrity. Online proctoring may occur through a variety of methods, including but not limited to:

a. live online proctoring where a remote invigilator authenticates identity and observes completion of an exam using specialized software and recordings;
b. automated proctoring where the exam session is recorded and AI (artificial intelligence) analyzed;
c. browser lockdown that limits access to other applications, websites, copying, printing, screen capture and other functions; or
d. a combination of both live/automated proctoring and browser lockdown.

Course instructors will review recordings, analyses, and data obtained through online proctoring for academic integrity infractions. It is the student's responsibility to meet the technical, software, location, and identity verification requirements necessary to enable online proctoring.

Further details of these policies are available in the Academic Calendar and on the Bow Valley College website, bowvalleycollege.ca.

Learners are encouraged to keep a copy of this course outline for future reference.

Collection of Personal Information:

This course, including your image and voice, may be recorded and made available to you and other students taking the course section. By attending the class(es) online or in person, you consent to the collection of your personal information. If you do not wish to be recorded, please contact your instructor before starting the course/class to discuss alternative arrangements.

You may use the recordings only for educational purposes and you must not copy, share, or use the recordings for any other purpose without the instructor's express permission.

Your personal information is collected in accordance with section 33(c) of the Freedom of Information and Protection of Privacy Act (Alberta) to deliver academic programming, support learner flexibility, promote universal design for learning principles, and for purposes consistent with the course activities and outcomes. If you have any questions about the collection, disclosure, use, or protection of this information, please contact the College's Access and Privacy Officer at foip@bowvalleycollege.ca.